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Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(C-C) = 0.003 \text{ Å}$ R factor = 0.049 wR factor = 0.124 Data-to-parameter ratio = 13.7

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1-Methoxyethyl-5-nitrobenzimidazole

The title compound, $C_{10}H_{11}N_3O_3$, was synthesized by the reaction of 5-nitrobenzimidazole, 2-chloroethyl methyl ether and KOH in ethanol. The bond lengths and angles are unexceptional.

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Comment

Nitrobenzimidazoles have attracted considerable interest due to their presence in a number of therapeutically and biologically active compounds. Nitrobenzimidazoles represent a group of nitro compounds that have seldom been characterized, although some of them possess antitrychomonal and other types of antimicrobial activities. Some nitrobenzimidazoles are relatively efficient substrates for DTdiaphorase and this enzyme is partly responsible for their cytotoxicity to bovine leukemia virus-transformed fibroblast culture (Sarlauskas *et al.*, 1997). We have also synthesized and nitrobenzimidazole derivatives (Öztürk *et al.*, 2003). We report here the synthesis of a biologically interesting nitrobenzimidazole compound; the results may be compared with those of a related heterocycle (Akkurt *et al.*, 2004).



The molecular structure with the atom-numbering scheme is shown in Fig. 1. The bond lengths and angles in (I) are within normal ranges (Allen *et al.*, 1987). The results agree with those for the benzimidazole and nitrobenzimidazole compounds (Akkurt, Karaca *et al.*, 2005; Akkurt, Türktekin *et al.*, 2005; Akkurt, Yıldırım Öztürk *et al.*, 2005). The benzimidazole ring is planar. The dihedral angle between the benzene and the fused five-membered ring system is 0.88 (11)°. The molecular packing is shown in Fig. 2.

Experimental

5-Nitrobenzimidazole (2.0 g, 12.26 mmol) and 2-chloroethyl methyl ether (2.3 ml, 25.18 mmol) were added to a solution of KOH (1.02 g, 18.39 mmol) in ethanol (25 ml) and the mixture was refluxed for 5 h. The precipitated KCl was then filtered off while the solution was hot. The solution was cooled to room temperature and a crude product precipitated. The precipitate was crystallized from ethanol (10 ml)

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(yield: 2.32 g, 77%; m.p. 351–352 K). ¹H NMR (CDCl₃): δ 3.4 (*s*, OCH₃, 3H), 3.8 (*t*, N–CH₂CH₂O, 2H), 4.4 (*t*, N–CH₂CH₂O, 2H), 7.2–8.4 (*m*, Ar–H, 3H), 8.7 (*s*, benzimidazole-C²–H, 1 H). Elemental analysis calculated for C₁₀H₁₁N₃O₃: C 54.29, H 4.98, N 19.00%; found: C 54.25, H 4.95, N 18.87%.

Crystal data

 $\begin{array}{l} C_{10}H_{11}N_3O_3 \\ M_r = 221.22 \\ \text{Triclinic, } P\overline{1} \\ a = 4.7940 \ (7) \ \mathring{A} \\ b = 8.4531 \ (13) \ \mathring{A} \\ c = 13.411 \ (2) \ \mathring{A} \\ \alpha = 97.350 \ (12)^{\circ} \\ \beta = 94.854 \ (12)^{\circ} \\ \gamma = 101.266 \ (12)^{\circ} \\ V = 525.25 \ (14) \ \mathring{A}^3 \end{array}$

Data collection

Stoe IPDS-II diffractometer
ω scans
Absorption correction: integration
(X-RED32; Stoe & Cie, 2002)
$T_{\min} = 0.920, \ T_{\max} = 0.995$
8616 measured reflections
2443 independent reflections

Refinement

Refinement on F^2
$R[F^2 > 2\sigma(F^2)] = 0.049$
$wR(F^2) = 0.124$
S = 0.98
2443 reflections
178 parameters

Z = 2 $D_x = 1.399 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 7061 reflections $\theta = 2.7-27.9^{\circ}$ $\mu = 0.11 \text{ mm}^{-1}$ T = 296 KPlate, colorless $0.80 \times 0.37 \times 0.05 \text{ mm}$

1327 reflections with $I > 2\sigma(I)$ $R_{int} = 0.049$ $\theta_{max} = 27.7^{\circ}$ $h = -6 \rightarrow 6$ $k = -11 \rightarrow 11$ $l = -17 \rightarrow 17$

H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.0607P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 0.14 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.20 \text{ e } \text{\AA}^{-3}$

H atoms were located in a difference synthesis and refined isotropically [C-H = 0.92 (3)–1.02 (4) Å]. The U_{iso} (H) values were set at 1.2 (1.5 for methyl group) times U_{eq} of the carrier atom.

Data collection: X-AREA (Stoe & Cie, 2002); cell refinement: X-AREA; data reduction: X-RED32 (Stoe & Cie, 2002); program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: ORTEP3 for Windows (Farrugia, 1997); software used to prepare material for publication: WinGX (Farrugia, 1999).

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An ORTEP3 drawing of (I), showing the atom-numbering scheme. Displacement ellipsoids are drawn at the 30% probability level.



Figure 2

The packing of molecules of (I) in the crystal structure.

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